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Ruthenium Hydride-Catalyzed Addition of Aldehydes to Dienes Leading to β , γ -Unsaturated Ketones

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Transition metal hydrides are involved in a variety of catalytic transformations, and hydrometalation of such species to unsaturated bonds provides intermediates having metal-carbon or heteroatom bonds.1 Ruthenium hydride catalysts have played a large role in these important transformations.^{2,3} We recently reported that the regioselective addition of aldehydes to unsaturated ketones leading to 1,3-diketones can be effectively catalyzed by RuHCl(CO)(PPh₃)₃ (eq 1).⁴ If a similarly regioselective hydroacylation takes place in the reaction between conjugate dienes and aldehydes, β , γ -unsaturated ketones would be formed (eq 2). A decade ago, employing Ru(cod)(cot)PPh3 as a catalyst, Kondo, Mitsudo and co-workers⁵ reported cross-condensation of conjugate dienes and aromatic aldehydes. Albeit pioneering efforts, their system suffers a narrow scope, including unavailability of aliphatic aldehydes and harsh reaction conditions with the use of dienes as a solvent. Considering the importance of this transformation and relying on the potential of ruthenium hydride complexes, we embarked on a study to explore an efficient catalytic system on a synthetically useful level. Herein we report that ruthenium hydride, RuHCl(CO)(PPh₃)₃,⁶ can serve as an efficient catalyst, providing a general method for the synthesis of of β , γ -unsaturated ketones from 1,3-dienes and aldehydes.



Using the reaction of isoprene (1a) and benzaldehyde (2a) as a model (Table 1), we surveyed ruthenium hydride catalysts. We were pleased to find that treatment of a toluene solution of 1a and 2a with RuHCl(CO)(PPh₃)₃ (5 mol %) at 90 °C for 24 h gave the desired cross-addition product, 2,3-dimethyl-1-phenyl-3-buten-1-one (3a) in 95% yield (entry 3). Other catalysts, such as RuH₂(CO)(PPh₃)₃ and RuHCl(PPh₃)₃, gave a smaller amount of the desired product 3a (entries 4 and 5).

Table 2 illustrates the wide generality and substrate scope of this transformation. Aryl aldehydes, such as *o*- and *p*-tolualdehydes, gave good yields of β , γ -unsaturated ketones **3b** and **3c**, respectively (entries 2 and 3). Similarly, *o*-, *m*-, and *p*-methoxybenzaldehydes (**2d**-**2f**) gave the corresponding ketones **3d**-**3f** in good yields (entries 4–6). *p*-Fluorobenzaldehyde (**2g**) also reacted with **1a** to give **3g** (entry 7). The reaction was effective for heteroaromatic aldehydes such as **2h** (entry 8) and aliphatic and α , β -unsaturated aldehydes **2i**, **2j**, and **2k** (entries 9–11). The reaction between **2g** and myrcene (**1b**), having one more double bond in the molecule, gave ketone **3l** in 53% yield (entry 12). Since the ruthenium hydride catalyst employed can affect double bond isomerization,^{4,6a,b,7} we tested nonconjugate diene **1c**, for which double bond migration would be followed by a cross-coupling reaction. Gratifyingly, **3m**

Table 1. Survey of RuH Catalysts^a

		RuH catalyst toluene, 90 °C, 24 h	Ja Ja
entry	1a (equiv)	catalyst (mol%)	yield of 3a (%) ^b
1	2	RuHCl(CO)(PPh ₃) ₃ (5)	65
2	2	RuHCl(CO)(PPh ₃) ₃ (10)	92
3	4	RuHCl(CO)(PPh ₃) ₃ (5)	95
4	4	RuH ₂ (CO)(PPh ₃) ₃ (5)	11
5	4	RuHCl(PPh ₃) ₃ (5)	3

^{*a*} Conditions: **1a** (2 or 4 mmol), **2a** (1 mmol), Ru catalyst (5 or 10 mol%), toluene (6 mL). Reaction was carried out in a screw capped test tube. ^{*b*} GC yields using hexadecane as an internal standard.

Scheme 1. Potential Mechanism



was obtained in good yield via the envisaged alkene-isomerization/ dehydrogenative carbonyl addition sequence (entry 13). In the reaction of *trans*-1,3-pentadiene (1d(E)) with 2a, α , β -unsaturated ketone **3n** was obtained as an E/Z mixture with a ratio of 27/73 (entry 14). Our RuH catalyst system also allows *cis* isomer 1d(Z)to react with 2a to give 3n (entry 15).⁸ The reaction of 4-methyl-1,3-pentadiene (1e) with 2a gave ketone 3o in 80% yield as a single product (entry 16). In the case of dialdehyde 2l, cascade-type crosscondensation took place to give lactone 3p albeit in moderate yield.

A possible mechanism for the present ruthenium hydride catalyzed reaction is illustrated by the reaction of **1a** with **2a** in Scheme 1. Addition of a ruthenium hydride to **1a** gives π -allyl-ruthenium complex **A**, as confirmed by ¹H NMR spectroscopy.^{9,10} No evidence for regioisomeric **B** was found in the NMR study, which may be due to steric considerations. The resulting complex **A** would then undergo reaction with aldehyde **2a** via a sixmembered transition state so as to place the bulky Ru-portion of the catalyst at the less hindered methylene carbon. The resulting

Table 2. Synthesis of β , γ -Unsaturated Ketones by a Ru–H Catalyzed Cross-Coupling Reaction of Dienes with Aldehydes^a



^a Conditions: 1 (4 equiv, 2 equiv for entries 4, 5, 6, 12, and 13), 2 (0.5 or 1 mmol), RuHCl(CO)(PPh₃)₃ (5 mol%), toluene (3 or 6 mL), 90 °C, 24 h. ^b Isolated yield by chromatography on SiO₂. ^c With 20 mol% of catalyst. ^d With 10 mol% catalyst. ^e NMR yield. ^f With 5 equiv of 1a. Isobenzofuranone was obtained in 32% yield.

complex C would then undergo a β -hydride elimination to give **3a** with liberation of ruthenium hydride. The formation of products 3n and 3o in the case of 1,3-pentadiene (1d (E,Z)) and 4-methyl-1,3-pentadiene (1e) (Table 2, entries 14, 15, and 16) may be rationalized by a similar pathway including isomerization of initially

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formed π -allylruthenium intermediate **D** to more stable Ru-complex E.¹¹ In a preceding study,⁵ however, Kondo et al. proposed a mechanism involving oxidative addition of benzaldehyde to Ru to form a benzoyl(hydride)ruthenium complex,12 which undergoes the consecutive hydroruthenation of dienes and reductive elimination. To determine whether their mechanism is operative in our RuH system, we carried out a crossover experiment using a mixture of PhCDO (2a-d) and p-FC₆H₄CHO (2g) with isoprene 1a. As the result, we observed nearly equal deuterium scrambling between products 3a-d and 3g-d. This result demonstrated that the hydrogen and acyl units that add to the diene do come from different molecules of aldehvdes.¹³

In summary, we have developed an efficient cross-addition reaction catalyzed by RuHCl(CO)(PPh₃)₃, which provides entry to a wide variety of β , γ -unsaturated ketones starting from the corresponding aldehydes and dienes. The detailed mechanism of this reaction as well as further extension of this chemistry is currently under investigation in this laboratory.

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Supporting Information Available: Experimental procedure, spectrum data of all products, NMR study. This material is available free of charge via Internet at http://pubs.acs.org.

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- (11) For similar isomerization of π -allylruthenium complexes, see ref 9b. (12) In our experiment, no change of chemical shift of carbonyl carbon was
- observed by NMR analysis when the Ru-H complex was treated with benzaldehvde.
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